## POZNAN UNIVERSITY OF TECHNOLOGY



## EUROPEAN CREDIT TRANSFER AND ACCUMULATION SYSTEM (ECTS)

pl. M. Skłodowskiej-Curie 5, 60-965 Poznań

## **COURSE DESCRIPTION CARD - SYLLABUS**

Course name

Podstawy technologii elektrochemiczne (Fundamentals of electrochemical technology)

Course

Field of study Year/Semester

Technologia chemiczna (Chemical Technology) 4/8

Area of study (specialization) Profile of study

general academic

Level of study Course offered in

First-cycle studies Polish

Form of study Requirements
part-time compulsory

**Number of hours** 

Lecture Laboratory classes Other (e.g. online)

20 20

Tutorials Projects/seminars

## **Number of credit points**

3

#### Lecturers

Responsible for the course/lecturer:

Responsible for the course/lecturer:

dr hab. Piotr Krawczyk, prof. PP

## **Prerequisites**

Student has a ordered knowledge of mathematics and physical chemistry and he also has ability to use the basic techniques in a laboratory scale.

# **Course objective**

The aim of the course is to familiarize students with an overview of technical electrochemistry methods and develop skills for their practical application.

## **Course-related learning outcomes**

Knowledge

- 1. The knowledge in the field of basics of electrochemical processes -[ K\_W03, K\_W08, K\_W10],
- 2. The knowledge in the field of various electrochemical technologies -[ K W12, K W13, K W15].

Skills

1. The student has the ability to plan the technological processes, the selection of measurement techniques, he also has ability to define the appearing chemical reactions and the yielded products – [K\_U16, K\_U18, K\_U20].

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#### Social competences

- 1. The student understands the need for self-study and improvement of their professional competence –[K\_K01],
- 2. Student can act and cooperate in the group –[K\_K03].

## Methods for verifying learning outcomes and assessment criteria

Learning outcomes presented above are verified as follows:

Laboratory assessment on the basis of the current work during the laboratory and the written tests.

The written test.

#### **Programme content**

- 1. The principles of electrochemical processes.
- 2. Electrodes balances.
- 3. The mechanisms of electrode processes.
- 4. The selected electrochemical processes used for synthesis of chemical compounds and environmental protection.
- 5. The selected issues in the field of generation, conversion and storage of electrical energy in chemical power sources.

## **Teaching methods**

Lecture, problem lecture, explanation, didactic discussion, classes, project method, laboratory exercises

## **Bibliography**

#### **Basic**

- 1. A. Kisza Elektrochemia cz. I i II (Jonika i Elektrodyka) WNT, W-wa, 2001,
- 2. R. Dylewski, W. Gniot, M. Gonet, Elektrochemia przemysłowa, Wyd. Politechniki Śląskiej, 1999,
- 3. A. Czerwiński, Ogniwa, akumulatory, baterie, WNT, W-wa, 1999,
- 4. A. Ciszewski, Technologia chemiczna. Procesy elektrochemiczne, Wyd. Politechniki Poznańskiej, 2008.

#### Additional

- 1. A.V. da Rosa, Fundamentals of Renewable Energy Processes, Elsevier/Academic Press, 1990,
- 2. H. Scholl, T. Błaszczyk, P. Krzyczmonik, Elektrochemia, Wyd. Uniwersytetu Łódzkiego, 1998.





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# Breakdown of average student's workload

	Hours	ECTS
Total workload	75	3,0
Classes requiring direct contact with the teacher	42	1,5
Student's own work (literature studies, preparation for laboratory	33	1,5
classes, preparation for tests) <sup>1</sup>		

1

 $<sup>^{\</sup>mbox{\scriptsize 1}}$  delete or add other activities as appropriate